

Manual of Petroleum Measurement Standards Chapter 11—Physical Properties Data

Section 2, Part 5—A Simplified Vapor Pressure
Correlation for Commercial NGAs

GPA 8117

SEPTEMBER 2007

REAFFIRMED, AUGUST 2017



AMERICAN PETROLEUM INSTITUTE



Manual of Petroleum Measurement Standards Chapter 11—Physical Properties Data

Section 2, Part 5—A Simplified Vapor Pressure
Correlation for Commercial NGLs

GPA 8117

Measurement Coordination

SEPTEMBER 2007

REAFFIRMED, AUGUST 2017

Prepared for

American Petroleum Institute
200 Massachusetts Avenue, NW, Suite 1100
Washington, DC 20001

GPA Midstream Association
6060 S. American Plaza, St. E, No. 700
Tulsa, OK 74135



AMERICAN PETROLEUM INSTITUTE



Foreword

The purpose of this procedure is to provide a simplified means of estimating equilibrium vapor pressures of various natural gas liquids (NGLs) from a knowledge of the fluid's relative density (60°F/60°F) and process temperature. The intended application of this procedure is to provide the values of P_e (equilibrium vapor pressure) required to determine the pressure effect contributions to volume correction factors as specified in the American Petroleum Institute *Manual of Petroleum Measurement Standards (MPMS)* Chapter 11.1-2004^[1] (which superseded Chapter 11.2.1-1984^[2]) and Chapter 11.2.2^[3]. It is realized that other equations of state are currently in use for specific custody transfer applications and that such methods will continue to be used and acceptable for both buyer and seller.

This procedure is applicable to four major classifications of petroleum fluid mixtures: commercial propanes, commercial butanes, natural gasolines, and light end fluids. The latter consists of EP mixes and high ethane content fluids. It covers the relative density range of 0.350 to 0.675 over a temperature range of -50°F through 140°F. This procedure is an extension of GPA Technical Publication TP-1 (1988)^[9]/API *MPMS* Addendum to Chapter 11.2.2-1994^[4] to include light end fluids in the relative density range of 0.350 to 0.490.

Variations from the computed vapor pressures to the actual values are to be expected because of the infinite number of possible compositions that can result in the same relative density product. Representative and extreme compositions were selected to develop the correlations, but it is realized that additional streams with compositions from among the infinite potential may well behave differently. This potential for variation is especially true at relative densities in the neighborhood of 0.500. For example, at a relative density of 0.505 the fluid could be propane or Y-grade mix, each having significantly different compositions and vapor pressure behaviors.

As is always the case in correlations published for custody transfer and settlement purposes, additional accuracy may be obtained by developing a modified correlation for certain specific applications if agreed to by all contracting parties. An equation to improve the accuracy of the generalized correlation at 100°F is also included.

It is important to note that the application of the correlations presented in this document to conditions of fluids not specified, will result in untested and unknown results which could contain significant errors.

Nothing contained in any API publication is to be construed as granting any right, by implication or otherwise, for the manufacture, sale, or use of any method, apparatus, or product covered by letters patent. Neither should anything contained in the publication be construed as insuring anyone against liability for infringement of letters patent.

This document was produced under API standardization procedures that ensure appropriate notification and participation in the developmental process and is

designated as an API standard. Questions concerning the interpretation of the content of this publication or comments and questions concerning the procedures under which this publication was developed should be directed in writing to the Director of Standards, American Petroleum Institute, 200 Massachusetts Avenue, NW, Suite 1100, Washington, DC 20001. Requests for permission to reproduce or translate all or any part of the material published herein should also be addressed to the director.

Generally, API standards are reviewed and revised, reaffirmed, or withdrawn at least every five years. A one-time extension of up to two years may be added to this review cycle. Status of the publication can be ascertained from the API Standards Department, telephone (202) 682-8000. A catalog of API publications and materials is published annually and updated quarterly by API, 200 Massachusetts Avenue, NW, Suite 1100, Washington, DC 20001.

Suggested revisions are invited and should be submitted to the Standards and Publications Department, API, 200 Massachusetts Avenue, NW, Suite 1100, Washington, DC 20001, standards@api.org.

API Special Notes

API publications necessarily address problems of a general nature. With respect to particular circumstances, local, state, and federal laws and regulations should be reviewed.

Neither API nor any of API's employees, subcontractors, consultants, committees, or other assignees make any warranty or representation, either express or implied, with respect to the accuracy, completeness, or usefulness of the information contained herein, or assume any liability or responsibility for any use, or the results of such use, of any information or process disclosed in this publication. Neither API nor any of API's employees, subcontractors, consultants, or other assignees represent that use of this publication would not infringe upon privately owned rights.

API publications may be used by anyone desiring to do so. This publication is an updated version of *MPMS Addendum to Chapter 11.2.2*. Previous editions of this publication were numbered *MPMS Addendum to Chapter 11.2.2*. Users of this standard should take efforts to ensure they are using the most current version of this publication. Every effort has been made by the Institute to assure the accuracy and reliability of the data contained in them; however, the Institute makes no representation, warranty, or guarantee in connection with this publication and hereby expressly disclaims any liability or responsibility for loss or damage resulting from its use or for the violation of any authorities having jurisdiction with which this publication may conflict.

API publications are published to facilitate the broad availability of proven, sound engineering and operating practices. These publications are not intended to obviate the need for applying sound engineering judgment regarding when and where these publications should be utilized. The formulation and publication of API publications is not intended in any way to inhibit anyone from using any other practices.

Any manufacturer marking equipment or materials in conformance with the marking requirements of an API standard is solely responsible for complying with all the applicable requirements of the standard. API does not represent, warrant, or guarantee that such products do in fact conform to the applicable API standard.

All rights reserved. No part of this work may be reproduced, stored in a retrieval system, or transmitted by any means, electronic, mechanical, photocopying, recording, or otherwise, without prior written permission from the publisher.

GPA Disclaimer

Neither the GPA nor any person acting on behalf of the GPA makes any warranty, guarantee, or representation, express or implied, with respect to the accuracy, completeness, or usefulness of the information contained in this report. The GPA hereby expressly disclaims any liability or responsibility for loss or damage resulting from the use of any apparatus, method, or process disclosed in this report; and for the infringement of any patent or the violation of any federal, state, or municipal law or regulation arising from the use of, any information, apparatus, method, or process disclosed in this report.

All rights reserved. No part of this work may be reproduced, stored in a retrieval system, or transmitted by any means, electronic, mechanical, photocopying, recording, or otherwise, without prior written permission from the publisher.

Copyright © 2007 American Petroleum Institute, GPA Midstream Association

Table of Contents

<i>Foreword</i> _____	<i>ii</i>
<i>API Special Notes</i> _____	<i>iii</i>
<i>ASTM Note</i> _____	<i>v</i>
<i>GPA Disclaimer</i> _____	<i>v</i>
<i>Table of Contents</i> _____	<i>v</i>
<i>List of Tables</i> _____	<i>iii</i>
<i>List of Figures</i> _____	<i>viii</i>
0 Implementation Guidelines _____	1
1 Background _____	1
2 Vapor Pressure Correlation for Commercial NGLs _____	2
3 Correlation Development _____	3
3.1 Propanes _____	7
3.1.1 Product Specifications _____	7
3.1.2 Data Collection _____	8
3.1.3 Data Regression _____	8
3.1.4 Error Analysis _____	8
3.2 Butanes _____	11
3.2.1 Product Specifications _____	11
3.2.2 Data Collection _____	11
3.2.3 Data Regression _____	12
3.2.4 Error Analysis _____	12
3.3 Natural Gasolines _____	15
3.3.1 Product Specifications _____	15
3.3.2 Data Collection _____	15
3.3.3 Data Regression _____	17
3.3.4 Error Analysis _____	17
3.4 Light Ends _____	19
3.4.1 Product Specifications _____	19
3.4.2 Data Collection and Validation _____	20
3.4.3 Data Regression _____	22
3.4.4 Error Analysis _____	25
4 Ad Hoc Improvement of the Correlation For Specific Situations _____	25
5 List of References _____	26

List of Tables

Table 1: Parameters for Vapor Pressure Correlation (Use in Equation 2)	3
Table 2: GPA Liquefied Petroleum Gas Specifications: GPA Standard 2140-88 ⁽⁷⁾	7
Table 3: Correlation Parameters for Propanes and Butanes	9
Table 4: SRK Interaction Parameters for Propanes and Butanes	9
Table 5: Compositions and Relative Densities of Propane Samples	9
Table 6: Comparison of Vapor Pressure Correlations for Commercial Propanes	10
Table 7: Compositions and Relative Densities of Butane Samples Data Regression	12
Table 8: Comparison of Vapor Pressure Correlations for Commercial Butanes	13
Table 9: GPA Standard 3132-84, “Natural Gasoline Specifications and Test Methods”	15
Table 10: Grades of Natural Gasoline as specified by the GPA	15
Table 11: Correlation Constants for Natural Gasolines	15
Table 12: SRK Interaction Parameters for Natural Gasolines	16
Table 13: Compositions and Relative Densities of Natural Gasolines	17
Table 14: Comparison of Vapor Pressure Correlations for Natural Gasolines	18
Table 15: Compositions of Components Used to Generate Data for Light Ends Correlation	21
Table 16: Representative Comparison of Vapor Pressures Obtained from HYSYS with those from NGLCALC	22
Table 17: Representative Comparison Between HYSYS SRK Vapor Pressures and Vapor Pressures from the Correlation for Light End Fluids	23

List of Figures

Figure 1: Vapor Pressures from Correlations	5
Figure 2: “A” Parameter (Equation 2 & Table 1)	6
Figure 3: “B” Parameter (Equation 2 & Table 1)	6
Figure 4: Maximum Temperature vs Relative Density	21

A Simplified Vapor Pressure Correlation for Commercial NGLs

0 Implementation Guidelines

This Revised Standard/Technical Publication is effective upon the date of publication and supersedes all previous revisions of the Standard/Technical Publication and API *MPMS* 11.2.2A/GPA TP-15. However, due to the nature of the changes in this Revised Standard/Technical Publication and the fact that it is or may be incorporated by reference in various regulations, it is recognized that guidance concerning an implementation period may be needed in order to avoid disruptions within the industry and ensure proper application. As a result, it is recommended that this Revised Standard/Technical Publication be utilized on all new and existing applications no later than TWO YEARS after the publication date. An application, for this purpose, is defined as the point where the calculation is applied.

Once the Revised Standard/Technical Publication is implemented in a particular application, the Previous Standard/Technical Publication will no longer be used in that application.

However, the use of API standards and ASTM and GPA technical publications remains voluntary and the decision on when to utilize a standard/technical publication is an issue that is subject to the negotiations between the parties involved in the transaction.

1 Background

The transfer of ownership of liquids is usually based on the volume of liquid at agreed upon standard conditions, usually 60°F for the U.S. customary system of units and the greater of one atmosphere pressure or the equilibrium vapor pressure of the liquid. Actual measurement of the liquid volumes and the their associated densities occurs at flowing or process conditions. Thus these measurements must be converted to equivalent values at the standard conditions. Once the liquid densities are converted, the conversion of the volumes becomes a trivial exercise. Densities are normally converted from measured conditions to standard conditions by equations of the form:

$$\gamma_{60} = F_t \times F_p \times \gamma \quad \text{Equation 1}$$

Where:

- γ_{60} Relative Density at 60°F and the greater of one atmosphere pressure or the equilibrium vapor pressure of the liquid
- γ Relative Density at measured conditions
- F_t Correction factor for temperature effects
- F_p Correction factor for pressure effects

Two methods used for calculation of the F_p term were standardized by the American Petroleum Institute: *MPMS* Chapter 11.2.1-1984^[2] (now superseded by Chapter 11.1-2004^[1]) and *MPMS* Chapter 11.2.2-1986^[3]. These methods require a knowledge of the equilibrium bubble point pressure (vapor pressure) at the measured conditions. However, the vapor pressure of the process liquid is generally not measured. The vapor pressure can also be calculated from compositional information, but the composition is not always measured for natural gas liquids, NGLs.